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**(E)-N'-(4-Chlorobenzylidene)-3,4,5-trimethoxybenzohydrazide**

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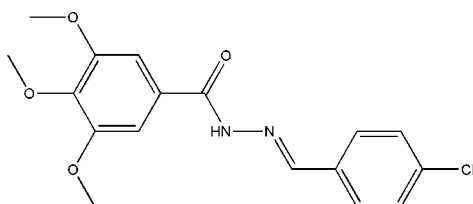
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Key indicators: single-crystal X-ray study;  $T = 273$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.046;  $wR$  factor = 0.128; data-to-parameter ratio = 13.1.

The title compound,  $\text{C}_{17}\text{H}_{17}\text{ClN}_2\text{O}_4$ , was synthesized from 3,4,5-trimethoxybenzohydrazide and 4-chlorobenzaldehyde. In the crystal structure, packing is stabilized by intramolecular  $\text{C}-\text{H}\cdots\text{O}$  and intermolecular  $\text{N}-\text{H}\cdots\text{O}$  and  $\text{C}-\text{H}\cdots\text{O}$  hydrogen-bonding interactions.

## Related literature

For related literature, see: Yang *et al.* (1996); Nawar *et al.* (2000); Gardner *et al.* (1991); Labouta *et al.* (1989); Wang *et al.* (2008); Allen *et al.* (1987).



## Experimental

## Crystal data

$\text{C}_{17}\text{H}_{17}\text{ClN}_2\text{O}_4$   
 $M_r = 348.78$   
 Triclinic,  $P\bar{1}$   
 $a = 5.119$  (2) Å  
 $b = 8.210$  (4) Å  
 $c = 20.276$  (9) Å

$\alpha = 101.055$  (7)°  
 $\beta = 92.362$  (7)°  
 $\gamma = 101.459$  (7)°  
 $V = 816.9$  (7) Å<sup>3</sup>  
 $Z = 2$   
 Mo  $K\alpha$  radiation

$\mu = 0.26$  mm<sup>-1</sup>  
 $T = 273$  (2) K

0.12 × 0.10 × 0.06 mm

## Data collection

Bruker APEX CCD area-detector diffractometer 4307 measured reflections  
 2860 independent reflections  
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996) 2497 reflections with  $I > 2\sigma(I)$   
 $T_{\min} = 0.970$ ,  $T_{\max} = 0.985$   $R_{\text{int}} = 0.017$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.046$  218 parameters  
 $wR(F^2) = 0.128$  H-atom parameters constrained  
 $S = 1.03$   $\Delta\rho_{\text{max}} = 0.59$  e Å<sup>-3</sup>  
 2860 reflections  $\Delta\rho_{\text{min}} = -0.39$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N2}-\text{H2}\cdots\text{O1}^{\text{i}}$	0.86	2.18	2.943 (3)	147
$\text{C8}-\text{H8C}\cdots\text{O4}$	0.96	2.26	2.896 (4)	123
$\text{C11}-\text{H11}\cdots\text{O1}^{\text{i}}$	0.93	2.43	3.145 (3)	134
$\text{C16}-\text{H16}\cdots\text{O1}^{\text{ii}}$	0.93	2.57	3.368 (3)	144

Symmetry codes: (i)  $x + 1, y, z$ ; (ii)  $-x + 1, -y, -z + 1$ .

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXTL.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: AT2683).

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**supplementary materials**

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## (*E*)-*N'*-(4-Chlorobenzylidene)-3,4,5-trimethoxybenzohydrazide

Y.-M. Wang, Z.-D. Zhao, Y.-X. Chen and L.-W. Bi

### Comment

3,4,5-Trimethoxybenzohydrazide and their deviates show moderate fungicidal and anti-bacterial activities (Gardner *et al.*, 1991). The antibacterial activity of formylhydrazines and formylhydrazones has been reported by Labouta *et al.* (1989). Many derivatives of formylhydrazines have interesting biological properties. So we synthesized several derivatives of 3,4,5-trimethoxybenzohydrazide. In our previous paper we have reported the crystal structure of (*E*)-*N'*-(2-hydroxybenzylidene)-3,4,5-trimethoxybenzohydrazide (Wang *et al.*, 2008). Now we synthesized the title compound (I) and report here its crystal structure.

The molecular structure of (I) is shown in Fig. 1. All bond lengths and angles in (I) are normal (Allen *et al.*, 1987). In the crystal structure, there exist intramolecular C—H $\cdots$ O, and intermolecular N—H $\cdots$ O and C—H $\cdots$ O hydrogen bonding interactions (Table 1, Fig. 2).

### Experimental

An ethanol solution (50 ml) of 3,4,5-trimethoxybenzohydrazide (0.01 mol) and 4-chlorobenzaldehyde (0.01 mol) was refluxed and stirred for 2 h; the mixture was cooled and the resulting solid product, (I), was collected by filtration. Crystals suitable for single-crystal X-ray diffraction were grown by slow evaporation of a solution in THF.

### Refinement

All H atoms were placed geometrically with C—H = 0.93–0.96 Å and N—H = 0.86 Å, and included in the refinement in riding motion approximation with  $U_{\text{iso}}(\text{H}) = 1.2$  or  $1.5U_{\text{eq}}$  of the carrier atom.

### Figures

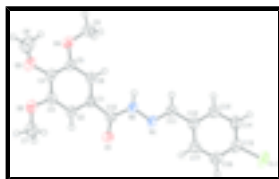


Fig. 1. A view of the molecular structure of (I), showing displacement ellipsoids at the 50% probability level.

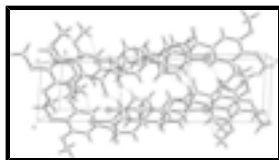


Fig. 2. Packing diagram of the title structure.

## (E)-N'-(4-Chlorobenzylidene)-3,4,5-trimethoxybenzohydrazide

### Crystal data

$C_{17}H_{17}ClN_2O_4$	$Z = 2$
$M_r = 348.78$	$F_{000} = 364$
Triclinic, $P\bar{1}$	$D_x = 1.418 \text{ Mg m}^{-3}$
Hall symbol: -P 1	Mo $K\alpha$ radiation
$a = 5.119 (2) \text{ \AA}$	$\lambda = 0.71073 \text{ \AA}$
$b = 8.210 (4) \text{ \AA}$	Cell parameters from 2557 reflections
$c = 20.276 (9) \text{ \AA}$	$\theta = 2.6\text{--}28.3^\circ$
$\alpha = 101.055 (7)^\circ$	$\mu = 0.26 \text{ mm}^{-1}$
$\beta = 92.362 (7)^\circ$	$T = 273 (2) \text{ K}$
$\gamma = 101.459 (7)^\circ$	Block, yellow
$V = 816.9 (7) \text{ \AA}^3$	$0.12 \times 0.10 \times 0.06 \text{ mm}$

### Data collection

Bruker APEX CCD area-detector diffractometer	2860 independent reflections
Radiation source: fine-focus sealed tube	2497 reflections with $I > 2\sigma(I)$
Monochromator: graphite	$R_{\text{int}} = 0.017$
$T = 273(2) \text{ K}$	$\theta_{\text{max}} = 25.0^\circ$
$\varphi$ and $\omega$ scans	$\theta_{\text{min}} = 2.1^\circ$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$h = -6 \rightarrow 6$
$T_{\text{min}} = 0.970$ , $T_{\text{max}} = 0.985$	$k = -9 \rightarrow 9$
4307 measured reflections	$l = -18 \rightarrow 24$

### Refinement

Refinement on $F^2$	Hydrogen site location: inferred from neighbouring sites
Least-squares matrix: full	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.046$	$w = 1/[\sigma^2(F_o^2) + (0.0602P)^2 + 0.4783P]$
$wR(F^2) = 0.128$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.03$	$(\Delta/\sigma)_{\text{max}} < 0.001$
2860 reflections	$\Delta\rho_{\text{max}} = 0.59 \text{ e \AA}^{-3}$
218 parameters	$\Delta\rho_{\text{min}} = -0.39 \text{ e \AA}^{-3}$
Primary atom site location: structure-invariant direct methods	Extinction correction: SHELXL97 (Sheldrick, 2008), $F_c^* = kF_c[1 + 0.001x F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$
Secondary atom site location: difference Fourier map	Extinction coefficient: 0.060 (5)

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C11	0.98386 (15)	-0.10593 (8)	0.26691 (3)	0.0637 (2)
O1	0.5118 (3)	0.4117 (2)	0.66278 (7)	0.0510 (4)
O2	0.5848 (4)	0.5136 (2)	0.92020 (8)	0.0634 (5)
O3	0.9795 (4)	0.7795 (3)	0.95374 (9)	0.0883 (7)
O4	1.3253 (4)	0.8754 (2)	0.86389 (9)	0.0703 (6)
N1	0.8782 (3)	0.3395 (2)	0.57654 (8)	0.0396 (4)
N2	0.9449 (3)	0.4260 (2)	0.64227 (8)	0.0393 (4)
H2	1.1099	0.4630	0.6572	0.047*
C1	0.8360 (4)	0.5329 (2)	0.75470 (9)	0.0365 (4)
C2	0.6692 (4)	0.4815 (3)	0.80214 (10)	0.0408 (5)
H2A	0.5169	0.3961	0.7888	0.049*
C3	0.7298 (4)	0.5574 (3)	0.86935 (10)	0.0457 (5)
C4	0.9504 (5)	0.6916 (3)	0.88879 (11)	0.0520 (6)
C5	1.1177 (4)	0.7419 (3)	0.84062 (11)	0.0480 (5)
C6	1.0627 (4)	0.6603 (3)	0.77344 (10)	0.0412 (5)
H6	1.1771	0.6909	0.7414	0.049*
C7	0.3631 (6)	0.3744 (4)	0.90386 (13)	0.0673 (7)
H7A	0.2345	0.3994	0.8736	0.101*
H7B	0.2821	0.3549	0.9443	0.101*
H7C	0.4225	0.2748	0.8827	0.101*
C8	1.2015 (8)	0.7925 (6)	0.99320 (16)	0.1086 (14)
H8A	1.2011	0.6860	1.0061	0.163*
H8B	1.2073	0.8791	1.0328	0.163*
H8C	1.3554	0.8220	0.9690	0.163*
C9	1.5185 (5)	0.9198 (3)	0.81913 (13)	0.0563 (6)
H9A	1.5826	0.8211	0.7990	0.084*
H9B	1.6651	1.0047	0.8435	0.084*
H9C	1.4392	0.9639	0.7845	0.084*
C10	0.7487 (4)	0.4520 (2)	0.68292 (10)	0.0363 (4)
C11	1.0749 (4)	0.3316 (3)	0.54076 (10)	0.0421 (5)
H11	1.2448	0.3915	0.5585	0.050*
C12	1.0395 (4)	0.2302 (3)	0.47225 (10)	0.0378 (4)
C13	1.2396 (5)	0.2560 (3)	0.42946 (11)	0.0527 (6)

## supplementary materials

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H13	1.3898	0.3424	0.4438	0.063*
C14	1.2205 (5)	0.1556 (3)	0.36576 (11)	0.0542 (6)
H14	1.3553	0.1749	0.3371	0.065*
C15	1.0009 (5)	0.0277 (3)	0.34555 (10)	0.0433 (5)
C16	0.7968 (5)	-0.0006 (3)	0.38638 (12)	0.0540 (6)
H16	0.6471	-0.0871	0.3717	0.065*
C17	0.8181 (4)	0.1019 (3)	0.44969 (11)	0.0488 (5)
H17	0.6803	0.0841	0.4777	0.059*

### Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Cl1	0.0910 (5)	0.0539 (4)	0.0397 (3)	0.0153 (3)	0.0094 (3)	-0.0077 (3)
O1	0.0337 (8)	0.0722 (11)	0.0394 (8)	0.0075 (7)	0.0002 (6)	-0.0034 (7)
O2	0.0650 (11)	0.0788 (12)	0.0361 (9)	-0.0015 (9)	0.0168 (7)	0.0009 (8)
O3	0.0691 (12)	0.1269 (19)	0.0400 (10)	-0.0058 (12)	0.0062 (9)	-0.0264 (11)
O4	0.0600 (11)	0.0767 (12)	0.0477 (10)	-0.0183 (9)	0.0098 (8)	-0.0190 (8)
N1	0.0417 (9)	0.0440 (9)	0.0293 (8)	0.0095 (7)	0.0003 (7)	-0.0018 (7)
N2	0.0339 (8)	0.0504 (10)	0.0288 (8)	0.0084 (7)	-0.0004 (6)	-0.0029 (7)
C1	0.0368 (10)	0.0400 (10)	0.0318 (10)	0.0128 (8)	0.0031 (8)	0.0002 (8)
C2	0.0386 (11)	0.0433 (11)	0.0361 (11)	0.0070 (9)	0.0043 (8)	-0.0009 (9)
C3	0.0458 (12)	0.0555 (13)	0.0341 (11)	0.0120 (10)	0.0102 (9)	0.0027 (9)
C4	0.0493 (12)	0.0641 (15)	0.0329 (11)	0.0084 (11)	0.0046 (9)	-0.0106 (10)
C5	0.0415 (11)	0.0534 (13)	0.0393 (11)	0.0042 (10)	0.0022 (9)	-0.0080 (10)
C6	0.0380 (11)	0.0472 (12)	0.0343 (10)	0.0081 (9)	0.0062 (8)	-0.0015 (9)
C7	0.0720 (17)	0.0708 (17)	0.0523 (15)	-0.0019 (14)	0.0221 (13)	0.0100 (13)
C8	0.092 (2)	0.168 (4)	0.0460 (17)	0.008 (2)	-0.0029 (16)	-0.005 (2)
C9	0.0467 (13)	0.0550 (14)	0.0567 (14)	-0.0008 (10)	0.0047 (11)	-0.0024 (11)
C10	0.0349 (10)	0.0385 (10)	0.0331 (10)	0.0072 (8)	0.0017 (8)	0.0020 (8)
C11	0.0402 (11)	0.0479 (12)	0.0333 (10)	0.0037 (9)	0.0022 (8)	0.0024 (9)
C12	0.0421 (11)	0.0393 (10)	0.0314 (10)	0.0108 (8)	0.0033 (8)	0.0037 (8)
C13	0.0483 (13)	0.0572 (14)	0.0417 (12)	-0.0043 (10)	0.0096 (10)	-0.0020 (10)
C14	0.0599 (14)	0.0586 (14)	0.0400 (12)	0.0062 (11)	0.0178 (10)	0.0033 (10)
C15	0.0584 (13)	0.0389 (11)	0.0329 (10)	0.0161 (9)	0.0037 (9)	0.0019 (8)
C16	0.0534 (13)	0.0481 (13)	0.0493 (13)	-0.0013 (10)	0.0046 (10)	-0.0054 (10)
C17	0.0457 (12)	0.0510 (13)	0.0425 (12)	0.0015 (10)	0.0118 (9)	-0.0013 (10)

### Geometric parameters ( $\text{\AA}$ , $^\circ$ )

Cl1—C15	1.742 (2)	C7—H7A	0.9600
O1—C10	1.223 (2)	C7—H7B	0.9600
O2—C3	1.357 (3)	C7—H7C	0.9600
O2—C7	1.419 (3)	C8—H8A	0.9600
O3—C8	1.337 (4)	C8—H8B	0.9600
O3—C4	1.362 (3)	C8—H8C	0.9600
O4—C5	1.361 (3)	C9—H9A	0.9600
O4—C9	1.411 (3)	C9—H9B	0.9600
N1—C11	1.270 (3)	C9—H9C	0.9600
N1—N2	1.379 (2)	C11—C12	1.460 (3)

N2—C10	1.349 (3)	C11—H11	0.9300
N2—H2	0.8600	C12—C17	1.379 (3)
C1—C2	1.383 (3)	C12—C13	1.380 (3)
C1—C6	1.383 (3)	C13—C14	1.381 (3)
C1—C10	1.489 (3)	C13—H13	0.9300
C2—C3	1.380 (3)	C14—C15	1.365 (3)
C2—H2A	0.9300	C14—H14	0.9300
C3—C4	1.395 (3)	C15—C16	1.371 (3)
C4—C5	1.393 (3)	C16—C17	1.381 (3)
C5—C6	1.388 (3)	C16—H16	0.9300
C6—H6	0.9300	C17—H17	0.9300
C3—O2—C7	117.98 (18)	O3—C8—H8C	109.5
C8—O3—C4	120.7 (3)	H8A—C8—H8C	109.5
C5—O4—C9	118.23 (18)	H8B—C8—H8C	109.5
C11—N1—N2	114.90 (17)	O4—C9—H9A	109.5
C10—N2—N1	119.35 (16)	O4—C9—H9B	109.5
C10—N2—H2	120.3	H9A—C9—H9B	109.5
N1—N2—H2	120.3	O4—C9—H9C	109.5
C2—C1—C6	121.06 (18)	H9A—C9—H9C	109.5
C2—C1—C10	116.43 (18)	H9B—C9—H9C	109.5
C6—C1—C10	122.40 (18)	O1—C10—N2	122.65 (18)
C3—C2—C1	119.68 (19)	O1—C10—C1	121.16 (17)
C3—C2—H2A	120.2	N2—C10—C1	116.20 (17)
C1—C2—H2A	120.2	N1—C11—C12	121.29 (19)
O2—C3—C2	124.6 (2)	N1—C11—H11	119.4
O2—C3—C4	115.37 (19)	C12—C11—H11	119.4
C2—C3—C4	120.01 (19)	C17—C12—C13	118.34 (19)
O3—C4—C5	122.0 (2)	C17—C12—C11	122.11 (18)
O3—C4—C3	118.0 (2)	C13—C12—C11	119.44 (19)
C5—C4—C3	119.73 (19)	C12—C13—C14	121.1 (2)
O4—C5—C6	124.2 (2)	C12—C13—H13	119.4
O4—C5—C4	115.72 (19)	C14—C13—H13	119.4
C6—C5—C4	120.0 (2)	C15—C14—C13	119.0 (2)
C1—C6—C5	119.33 (19)	C15—C14—H14	120.5
C1—C6—H6	120.3	C13—C14—H14	120.5
C5—C6—H6	120.3	C14—C15—C16	121.5 (2)
O2—C7—H7A	109.5	C14—C15—C11	119.15 (17)
O2—C7—H7B	109.5	C16—C15—C11	119.30 (17)
H7A—C7—H7B	109.5	C15—C16—C17	118.7 (2)
O2—C7—H7C	109.5	C15—C16—H16	120.7
H7A—C7—H7C	109.5	C17—C16—H16	120.7
H7B—C7—H7C	109.5	C12—C17—C16	121.3 (2)
O3—C8—H8A	109.5	C12—C17—H17	119.3
O3—C8—H8B	109.5	C16—C17—H17	119.3
H8A—C8—H8B	109.5		
C11—N1—N2—C10	175.22 (19)	O4—C5—C6—C1	-175.8 (2)
C6—C1—C2—C3	-0.3 (3)	C4—C5—C6—C1	2.2 (3)
C10—C1—C2—C3	-176.58 (19)	N1—N2—C10—O1	-5.0 (3)

## supplementary materials

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C7—O2—C3—C2	3.4 (4)	N1—N2—C10—C1	174.86 (16)
C7—O2—C3—C4	-178.0 (2)	C2—C1—C10—O1	33.9 (3)
C1—C2—C3—O2	-178.0 (2)	C6—C1—C10—O1	-142.3 (2)
C1—C2—C3—C4	3.4 (3)	C2—C1—C10—N2	-145.93 (19)
C8—O3—C4—C5	-63.6 (4)	C6—C1—C10—N2	37.9 (3)
C8—O3—C4—C3	122.7 (3)	N2—N1—C11—C12	173.85 (18)
O2—C3—C4—O3	-8.6 (3)	N1—C11—C12—C17	-19.4 (3)
C2—C3—C4—O3	170.1 (2)	N1—C11—C12—C13	164.4 (2)
O2—C3—C4—C5	177.6 (2)	C17—C12—C13—C14	-0.4 (4)
C2—C3—C4—C5	-3.7 (4)	C11—C12—C13—C14	176.0 (2)
C9—O4—C5—C6	-9.0 (4)	C12—C13—C14—C15	-0.8 (4)
C9—O4—C5—C4	172.9 (2)	C13—C14—C15—C16	1.5 (4)
O3—C4—C5—O4	5.4 (4)	C13—C14—C15—C11	-176.79 (19)
C3—C4—C5—O4	179.0 (2)	C14—C15—C16—C17	-1.0 (4)
O3—C4—C5—C6	-172.7 (2)	C11—C15—C16—C17	177.36 (19)
C3—C4—C5—C6	0.9 (4)	C13—C12—C17—C16	1.0 (3)
C2—C1—C6—C5	-2.5 (3)	C11—C12—C17—C16	-175.3 (2)
C10—C1—C6—C5	173.5 (2)	C15—C16—C17—C12	-0.3 (4)

### Hydrogen-bond geometry ( $\text{\AA}$ , $^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N2—H2 $\cdots$ O1 <sup>i</sup>	0.86	2.18	2.943 (3)	147
C8—H8C $\cdots$ O4	0.96	2.26	2.896 (4)	123
C11—H11 $\cdots$ O1 <sup>i</sup>	0.93	2.43	3.145 (3)	134
C16—H16 $\cdots$ O1 <sup>ii</sup>	0.93	2.57	3.368 (3)	144

Symmetry codes: (i)  $x+1, y, z$ ; (ii)  $-x+1, -y, -z+1$ .



Fig. 1

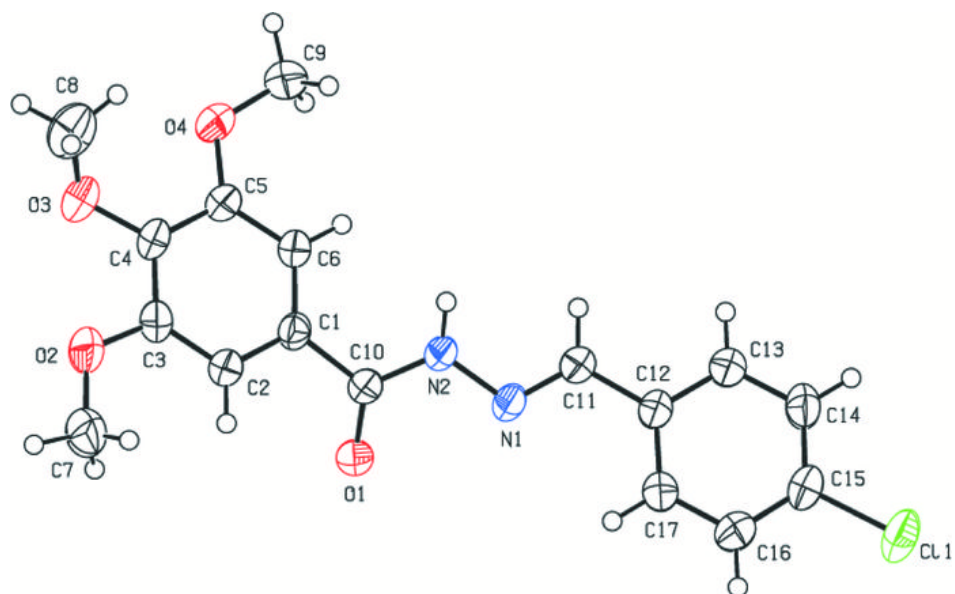


Fig. 2

